Brooklyn College Chemistry 1007 Fall 2015 Syllabus - Davenport Sections M9AL-M9EL

**Course Goals and Learning Objectives.** The goal of this course is to give the student a basic understanding of chemistry and physical processes in the context of environmental and food chemistry. The course also necessarily makes connections to the future sustainability of food and water. We aim to prepare the student for the increasingly urgent and complex national dialogue on the interrelated topics of global climate change, energy, pollution, extinction and the food supply. The specific objectives of this course are to provide the student with the basic vocabulary of chemistry, and a basic understanding of the experimental process as it relates to environmental and food chemistry.

#### Required Texts:

 "Chemistry in Context, 8<sup>th</sup> edition, McGraw Hill. The e-book is available through ConnectPlus (~\$85.00) and access for the first TWO weeks is FREE after you have registered (see below). For an additional \$15.00 you can also obtain the hardcopy version of the book.

You can register for ConnectPlus by following these steps:

- a. Make sure your pop-up blocker is 'OFF.'
- b. Go to the URL for your lab instructor/section: <u>http://connect.mheducation.com/class/davenp</u>
- ort-m9af-belyayeva-lab-instructor http://connect.mheducation.com/class/davenp
- ort-m9bf-belyayeva-lab-instructor

<u>http://connect.mheducation.com/class/davenp</u> ort-m9cb-elie-lab-instructor

<u>http://connect.mheducation.com/class/davenp</u> ort-m9db-buzin-lab-instructor

<u>http://connect.mheducation.com/class/davenp</u> ort-m9eb-lee-lab-instructor

- c. Click the "Register Now" Button.
- d. Enter your email address.
- e. Select "Buy Online" and purchase ConnectPlus.

2. "Experiments for Core Chemistry at Brooklyn College" (ISBN 9781465270191), \$32.80.

**Also required:** A small combination or key lock to secure lab supplies in the lab desk cabinet.

NB: LAB ATTENDANCE IS NOT OPTIONAL.

**Note**: Online (graded) problem sets are assigned through Blackboard. Please ensure that you have access to this class through Blackboard, and check that your posted email address is the one that you access regularly.

# **Class organization**

The order of the class *reading material* from ConnectPlus/Chemistry in Context is given below. However, the instructor reserves the right to make adjustments when necessary. Students may choose to use the ConnectPlus reading option, which allows for student query as reading is carried out. Otherwise, students can click on the e-book cover and 'open' the text for reading in the conventional way. Online problem sets will be assigned through Blackboard; these are graded online.

# Chapter 0

Sustainability, renewable and nonrenewable resources 0.1 - 0.3

Footprints: ecological, carbon, water, air - how our food choices impact these 0.5; Chapter 1: Intro

# Chapter 1

Necessary basics: classification of matter, the Periodic Table 1.1, 1.6- 1.7

Naming compounds; combustion 1.8 – 1.9, 1.10 (pp. 39 -40 only)

# Chapter 2

Atomic structure, periodicity, molecules and models 2.2 - 2.3

Light and color; biological effects of UV radiation: mitigation by micronutrients 2.4 & 2.7

# Chapter 5

The unique properties of water; hydrogen bonding Intro, 5.1 - 5.2

Potable water: future water sources, influence on food availability 5.3 - 5.4

Aqueous solutions, solutes, ionic compounds, naming ionic compounds 5.5 - 5.7

Solubility of Ionic Compounds 5.8

Covalent compounds and solutions, water purification 5.9, 5.11, pp. 237-38

MIDTERM: AFTER COMPLETION OF CHAPTER 5. APPROXIMATE DATE: Oct. 26<sup>th</sup> (in class)

# Chapter 6

What is an acid? What is a base? Intro, 6.1 - 6.2

Neutralization, introducing pH 6.3 - 6.4

The nitrogen cycle: source of amino acids for proteins, 6.9

# Chapter 9

Natural polymers, addition polymerization 9.2 - 9.3 Condensation polymerization; polyamides 9.6 - 9.7

# Chapter 11

Sustainability; malnourishment Intro, 11.1 - 11.2

Fats and oils 11.3

Trans vs. cis fats; interesterification 11.3

Kitchen as Lab: Chapter 10 To Bloom or Not to Bloom?

Carbohydrates and sugars 11.5 - 11.6

Kitchen as Lab: Chapter 24 Sweet Physics: Sugar, Sugar Blends and Sugar Glasses

Proteins 11.7

Vitamins and Minerals 11.8

Metabolism: energy from food; diet (quality vs. quantity 11.9 - 11.10

Locavorism; flexitarian, vegetarian, vegan 11.11 - 11.12

# Chapter 12

The structure of deoxyribonucleic acid 12.2 - 12.3

DNA: the code for proteins 12.4 - 12.5

Genetic Engineering 12.6 - 12.8

12/21: (Monday) FINAL EXAM, 8:30 am -10:00 am, Whitman Hall auditorium (bring pencils)

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#### Instructor Contact Information

Prof. Davenport	2825	344NE
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LDvnport@brooklyn.cuny.edu Office hours: 11:00am - 12:30pm Monday 11:00am - 12:30pm Wednesday or by appointment (email first).

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Course Requirements and Grading The final grade for the course is based on a score, which is the sum of the scores received for the following:

The grade breakdown is as follows:

**30%** lab grade <sup>1</sup>

15% online problem sets (graded)

25% midterm exam

30% final exam grade

EXAMS Questions from the lab experiments may be included in both the midterm and final exam. Calculators or use of other electronic devices are not necessary nor are they allowed during exams but you will need pencils.

The final exam for is scheduled for Monday, December 21<sup>st</sup>, 2015 from 8:30 am - 10:00 am in Whitman Hall auditorium. Bring pencils. The final exam is cumulative and will cover course material both covered and not covered by the midterm. Questions related to lab experiments may also appear on exams. <u>Bring pencils</u>!!! Be sure to arrive on time. Exams will not be available after the first student leaves OR after 9:00 am (whichever comes first).

It is the student's responsibility to note the midterm and final exam time, place and date at the beginning of the semester, and to be sure not to schedule other activities during this time. There is no make-up mid-term exam. If the midterm exam is missed, the final exam will be count towards both the midterm and final exam grade points.

Course work cannot be completed independent of the lab work. NO credit is earned for coursework without completion of the lab assignments.

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<sup>1</sup>Laboratory Schedule

NOTE: Labs meet EVERY WEEK.

<u>Lab meeting 1</u>: Check-in. lab safety, lab techniques. Intro to Experiment 1.

<u>Lab meeting 2 & 3:</u> Examination of the Physical and Chemical Properties of Matter

<u>Lab meeting 4:</u> Energy Powers Physical and Chemical Changes

<u>Lab meeting 5:</u> Name that Ion: Qualitative Analysis

<u>Lab meeting 6:</u> Water Makes the Difference -Decomposition of Copper Sulfate Pentahydrate

<u>Lab meeting 7:</u> Accounting for Every Atom: Moles in Chemical Reactions

<u>Lab meeting 8:</u> Carbon Dioxide: An all-too-common reaction product

<u>Lab meeting 9:</u> Bonding in Molecules: How Electrons control Physical and Chemical Properties

<u>Lab meeting 10 & 11:</u> An Experiment in 'Cleaning' Water

<u>Lab meeting 12:</u> Ester Synthesis: An Experiment that Smells Good

Lab meeting 13: Building Molecules with Models

Lab meeting 14: Check-out

Details about lab attendance, lab reports and grades. Attendance will be taken at the beginning of each lab class. It is the student's responsibility to notify the instructor of her/his presence.

Students are expected to come to lab ON TIME and be prepared by having read and understood the lab procedure BEFORE carrying out the work in class.

Each of the 10 lab report sheets must be handed in at the laboratory class meeting one week following the completion of the experiment. Lab report sheets are expected to be stapled together, and the <u>student's and lab instructor's</u> <u>name</u> must appear on each sheet. 10 points are awarded for each lab report. Point assignment is at the discretion of the lab instructor, but no less than 4 points shall be awarded for completion of the lab work.

Instructors deduct 3 points for each week the lab report is handed in late. Labs missed due to illness must be made up by attending alternate lab sessions with permission of the lab instructor, through the General arranged Chemistry stockroom technicians, Ms. Grace Kosiorek or Dr. Olga Berezovska (248 IE). Documentation of illness must be presented to the stockroom completed technician. Labs 50 must be accompanied by a signed lab form, obtained from the stockroom technician. This form is to be returned to the student's assigned lab instructor. Experiments not completed will result in a grade of zero for that experiment. LAB ATTENDANCE IS NOT OPTIONAL. STUDENTS WTI.I. RECEIVE A GRADE OF F FOR THE COURSE IF MORE THAN TWO LABS ARE MISSED AND NOT MADE UP. MISSED LABS MUST BE MADE UP WITHIN 2 WEEKS OF THE EXPERIMENT ASSIGNMENT. LAB MAKE-UPS ARE SUBJECT TO LAB SPACE AVAILABILITY.

Students are to work individually in the laboratory unless specifically told otherwise. Students are expected to actively participate in the collection of all data. 'Sharing' of results without actual participation in collection of those results constitutes cheating; no credit will be given for that lab. It is a New York State law that safety goggles must be worn at all times by all students in the laboratory. Goggles are provided as part of the lab equipment rental fee. Students who consistently refuse to properly wear safety goggles during the lab period will be dismissed from the laboratory. Students so dismissed will not have the opportunity to make up missed lab work. It is the student's responsibility to bring her/his goggles to each lab session. If the student completes their lab work before the end of the session, students MUST continue to wear their goggles until they have exited the lab.

Eating and drinking are also not permitted during lab sessions. Students may also be dismissed for violation of this safety rule.

Pregnant students are encouraged to defer taking chemistry 1007 as the lab is an integral part of the course.

Accommodations for Students with Disabilities In order to receive disability-related academic accommodations, students first must be registered with the **Center** Student for Disability Services. Students who have a documented disability or suspect they may have a disability are invited to set up an appointment with the Director of the Center for Student Disability Services, Ms. Valerie Stewart-Lovell, at 718-951-5538. If you have already registered with the Center for Student Disability Services, please professor provide your with the course accommodation form and discuss your specific accommodation with him/her.

# Academic Integrity

The faculty and administration of Brooklyn College support an environment free from cheating and plagiarism. Each student is responsible for being aware of what constitutes cheating and plagiarism and for avoiding both. The complete text of the CUNY Academic Integrity Policy and the Brooklyn College procedure for implementing that policy can be found at this site: <u>http://www.brooklyn.cuny.edu/bc/policies</u>. If a faculty member suspects a violation of academic integrity and, upon investigation, confirms that violation, or if the student admits the violation, the faculty member MUST report the violation.

All students should read carefully and thoroughly the 2014-2015 Brooklyn College Bulletin (http://www.brooklyn.cuny.edu/web/off\_registrar /141024\_2014-15\_Undergraduate\_Bulletin.pdf) for a complete listing of academic regulations of the College.

#### Classroom Etiquette

Eating and cell phone usage is not allowed during lecture. Unnecessary and superfluous movement, talking and other disruptive behavior distracts other students' attention from the lecture material to which they are entitled. Students who create such disturbances will be asked to leave and/or escorted from the classroom.

The state law regarding non-attendance because of **religious beliefs** shall be followed as given in the Brooklyn College Bulletin, Undergraduate Programs 2014-2015, p. 72 <u>http://www.brooklyn.cuny.edu/web/off\_registrar/</u> 141024\_2014-15\_Undergraduate\_Bulletin.pdf

#### **Important Dates**

Wednesday, September 2: Last day to add a course

Thursday, September 10: Conversion Day. Classes follow a Monday schedule.

Monday, November 9: Last day to withdraw from a course with a W (non-penalty) grade.