Brooklyn College Department of Chemistry

GENERAL CHEMISTRY IA, CHEM 1050 –Fall 2017

Required Text: Chemistry, The Central Science, Brown, LeMay, Bursten, and Murphy,

Pearson Pub., 2017, 14th Edition.

Required Items: Scientific calculator: Texas Instrument TI-30X recommended.

Graphing calculators and internet devices are not allowed on exams!

Online Supplements and Info:

www.brooklyn.cuny.edu/web/aca_naturalsciences_chemistry/Courses_Chem1100-Spr15-Syllabus.pdf (online **syllabus**)

http://academic.brooklyn.cuny.edu/chem/howell/practice.htm (old BC chemistry **exams**)

http://www.brooklyn.cuny.edu/web/academics/schools/naturalsciences/undergraduate/chemistry.php (Chemistry Department Homepage)

http://www.brooklyn.cuny.edu/web/academics/honors/prehealth.php (Pre-Health Professions website)

http://www.brooklyn.cuny.edu/web/aca_honors/131125_PrehealthProfessionsHandbook.pdf http://userhome.brooklyn.cuny.edu/mkobrak/labvideos.html (Lab instruction videos)

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Academic dishonesty is prohibited in the City University of New York.

Cheating, plagiarism, internet plagiarism and obtaining unfair advantages are violations of policies of academic integrity and are punishable by penalties, failing grades, suspension and expulsion. For more information about CUNY policy on academic integrity see http://www.brooklyn.cuny.edu/bc/policies/pdt7CUNY%20PolicyAcademicIntegrity.pdf

Student Disability Services

In order to receive disability-related academic accommodations students must first be registered with the Center for Student Disability Services. Students who have a documented disability or suspect they may have a disability are invited to set up an appointment with the Director of the Center for Student Disability Services, Ms. Valerie Stewart-Lovell at 718-951-5538. If you have already registered with the Center for Student Disability Services please provide your professor with the course accommodation form and discuss your specific accommodation with him/her.

LECTURE TESTS:

Note that these are during common hours.

FIRST TEST: **Tuesday, October 3, 12:30 – 2:00 PM**,

SECOND TEST: Tuesday, November 14, 12:30 – 2:00 PM,

NO Makeup exams are given for Lecture Tests. We mean it.

FINAL EXAM: MONDAY, DECEMBER 18, 8:00 AM – 10:00 AM

GRADING:

Your final grade will be determined as follows:

40% Two lecture tests

30% Quizzes: Dates to be determined by the lecturer

30% Final Exam

Drop/Add Dates:

August 24 Last day to drop a course with 0% liability (100% refund)

August 31 Last day to add a course;

August 31 Last day to drop a course with 25% liability.

September 7 Last day to drop a course with 50% liability.

September 14 Last day to drop a course without a grade of "W"

September 14 Last day to drop a course with 75% liability.

September 15 Course withdrawal period begins. A grade of "W" is assigned to

students who officially withdraw from a course.

November 10 Course withdrawal period ends.

Last day to withdraw from a class with a grade of "W".

Chem 1050 Assigned Reading

Below is the assigned reading and a corresponding set of homework problems. Your lecturer will give you guidance about where you are in the text and what to do to stay current with the reading. **Read the material at least once before the lecture**, and spend some time on the inchapter problems to reinforce it. Unless noted otherwise, problems listed as Homework correspond to the end-of-chapter problems for the corresponding chapter. Answers to red problems are at the end of the Chapter. **If you are instructed to memorize something, the test will be written assuming you have done so.**

Homework is assigned but not graded. Quiz and examination questions will mostly be similar to those given in the text. You should do as many of these as possible.

Assigned Reading	Topics	Homework
Appendix A.1	Math Review, Dimensional	Memorize Table 1.1 (in Ch.1)
Chapter 1	Analysis, Basic Concepts	Chapter 1: 1, 9, 13, 15, 16, 17, 19, 21, 24,
		33c, 34d, 35, 36a, 43, 45, 47, 49, 53, 55,
		57a,b,d,f, 83
Chapter 2	Elements, Compounds, Ions,	Memorize Tables 2.4 and 2.5
	Periodic Table	Chapter 2: 11, 13, 18, 27, 29, 33, 39, 41,
		49, 51, 55, 57, 58, 61, 65, 67 71, 73, 75,
		77, 81, 83, 111
Chapter 3	Chemical Equations, Moles,	Chapter 3: 1, 7, 9, 11, 15, 19, 21, 23,
	Empirical Formulas,	25a,b, 27, 30, 35, 37, 41, 45, 47, 51, 53,
	Stoichiometry, Limiting	55, 59, 61, 62, 80, 83, 86, 109
	Reagents	
Chapter 4	Chemical Reactions, Molarity,	Chapter 4: 14, 17, 21, 23,26, 33, 34, 37,
	Solution Stoichiometry	39, 45, 47, 49, 51, 53, 55, 59, 60, 61, 62,
		67, 71, 72, 73, 81, 83, 85
Chapter 5	Thermochemistry	Chapter 5: 19, 21, 23, 24, 25, 39, 43, 46,
		47, 51, 53, 55, 57, 62, 65, 67, 69, 73, 76,
		83, 91a